



# Ideas Hybrid Work

**IM Harris**



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$$\text{Pb(NO}_3)_2(\text{aq}) + 2\text{HCl(aq)} \rightarrow \text{PbCl}_2(\text{s}) + 2\text{HNO}_3$$

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web the net ionic equation for equation 1 is given as equation 3  $\text{s}^{2-} \text{ aq}$   $\text{Fe}^{2+} \text{ aq}$   $\text{gt}$   $\text{FeS} \text{ s}$  eq 3 8 note that the phase indicator aq is optional on ionic species but is

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