

Phase Equilibria for Carbon Dioxide + 1-Pentanol System at High Pressures



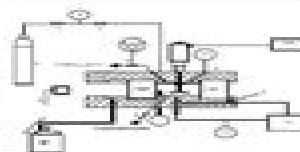
Cătălina Seculianu, Viorel Feroliu, and Dan Geană*

Politehnica University of Bucharest, Department of Applied Physical Chemistry & Electrochemistry
1-7 G-ral Gh. Polizu, St, 011061 Bucharest, ROMANIA

Introduction

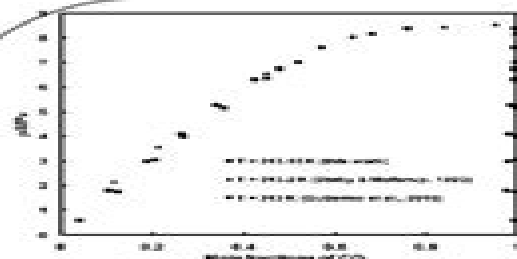
Experimental work

Phase behaviour measurements were made in a high-pressure visual cell with variable volume based on the static-analytical method [1,2].

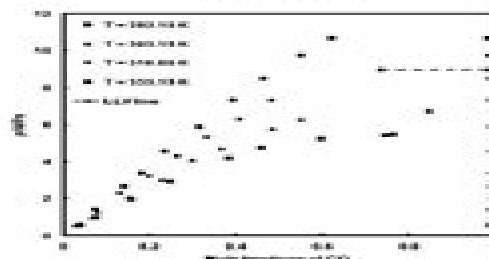


Vapour-liquid and vapour-liquid-liquid equilibria data for the carbon dioxide + 1-pentanol system at 293.15, 303.15, 316.65, and 333.15 K up to 10.67 MPa were determined. The higher temperatures branch of the three-phase liquid-liquid-vapour line and the upper critical endpoint and the lower critical endpoint were measured.

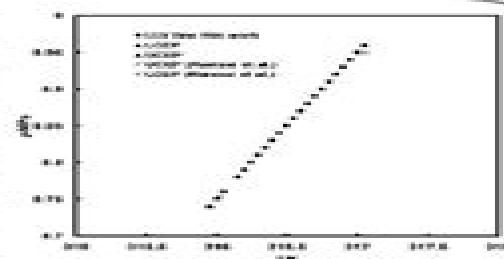
Results



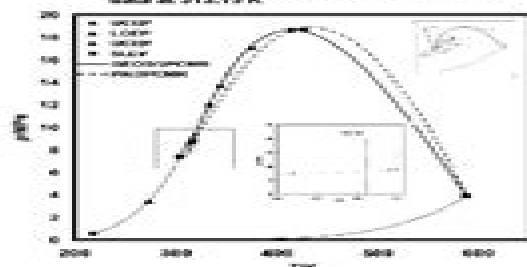
Comparison of measured and literature data at 313.15 K



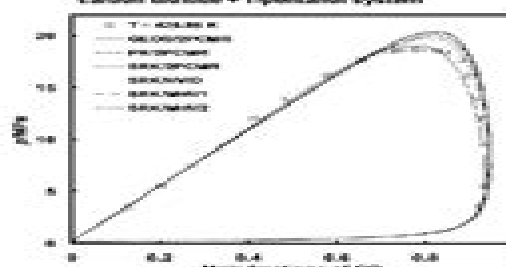
Pressure-composition data for carbon dioxide + 1-pentanol system



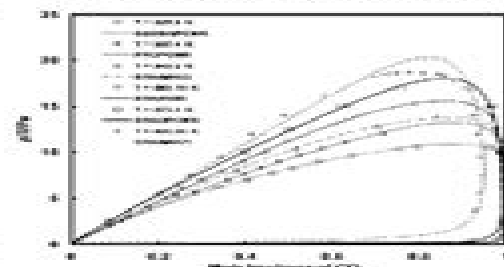
P-T projection of the LLV line (second branch)



P-T fluid phase diagram of carbon dioxide + 1-pentanol system



Comparison of literature VLE data at 325.65 K with model results



Comparison of measured and literature VLE data at different temperatures with model results

Acknowledgements

The authors are grateful to the National Council of Scientific Research of Romania (CNCS) for financial support.

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*Current address: Solar Carbonates & Carbon Storage Research Centre, Department of Chemical Engineering Imperial College London, South Kensington Campus, Exhibition Road, London SW7 2AZ U.K.
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