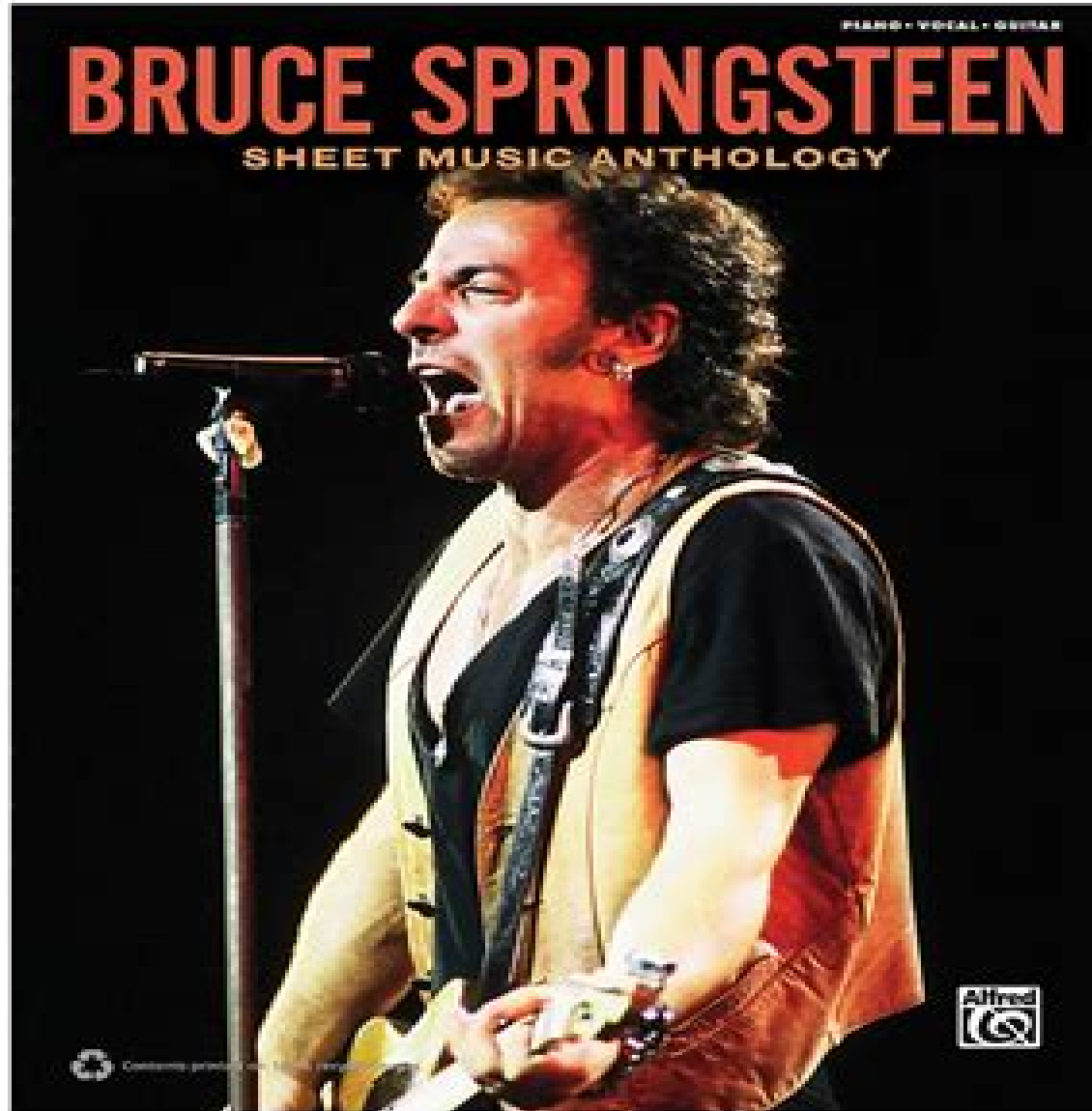


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web reacting two solutions eg acid and alkali place the polystyrene cup inside the glass beaker to make it more stable
measure an appropriate volume of each liquid eg 25 cm³ place one of the

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web for reactions involving acetic acid or ammonia the measured enthalpy change of neutralization is a few kJ less
exothermic than with strong acids and bases for example one source which gives the enthalpy change of neutralization of
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eg 1 hcl aq + naoh aq → nacl aq + h₂O l for which ΔH = -57 kJ mol⁻¹ eg 2 hNO₃ aq + KOH aq → KNO₃ aq + H₂O l for which ΔH = -57 kJ mol⁻¹

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