

BIOCHEMISTRY SUMMARY

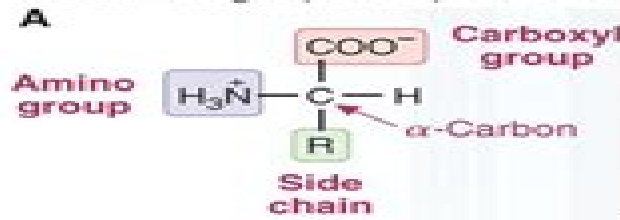
See written notes for I.1 and I.2.

I.3 – protein structure

Amino acids

= are the building blocks of proteins

- Contain an amino (NH_3^+) group and a carboxylic acid (COOH) group
- Both groups are connected to a central carbon atom called the alpha-carbon
- The other two groups are a proton and a side chain



- Acidity
 - Both the amino group and the carboxylic group are acidic – they can release a proton
 - At low pH (<2), amino acids are positively charged
 - At high pH ($>9-10$), amino acids are negatively charged
 - At intermediate pH (3-8), amino acids are zwitter ionic.
 - Zwitter ionic means that they contain one positive and one negative charge and do not have a net charge.

Stereo chemistry

= discipline that deals with chiral carbons because you need stereo view (3D)

- A carbon atom that contains four different chemical groups is chiral
 - It can occur in two mirror images. One cannot be turned into the other by rotation (i.e., many sugars and amino acids)
- The two mirror images are called enantiomers – they are historically indicated by D and L
 - The alpha-carbon of amino acids is chiral: it has four different chemical groups attached
 - It is common to display chiral molecules as shown here. Starting from the central chiral carbon atom, the bonds that are black and become thicker, point out of the screen/paper; bonds in grey (or dashed) that become thinner, point backward
 - Mammalian cells only use L-amino acids, D-amino acids are only rarely used by some bacteria, usually in antibiotics, because they are poorly degraded by most enzymes



Biochemistry A Synopsis

N Noddings



Biochemistry A Synopsis:

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